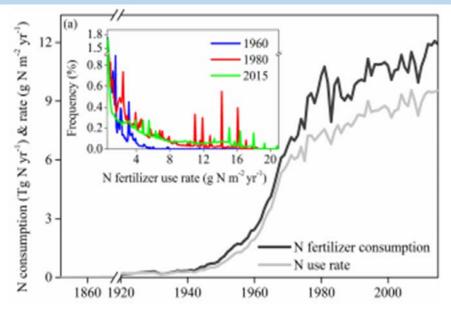
Comparison of storm-event transportation trends on Nitrate in Chloride in a low-gradient stream within Central Illinois

By Jackson Wassik

Nitrate (NO_3^-)





- NO₃⁻ is sourced from fertilizers whose usage has increased over the past 80 years^[1]
- With ever growing food demands, we can infer that NO₃⁻ usage will continue to increase.
- High concentrations of NO₃⁻ in surface waters are exacerbated by tile drainage, preventing subsurface flow and NO₃⁻ removal^[2]

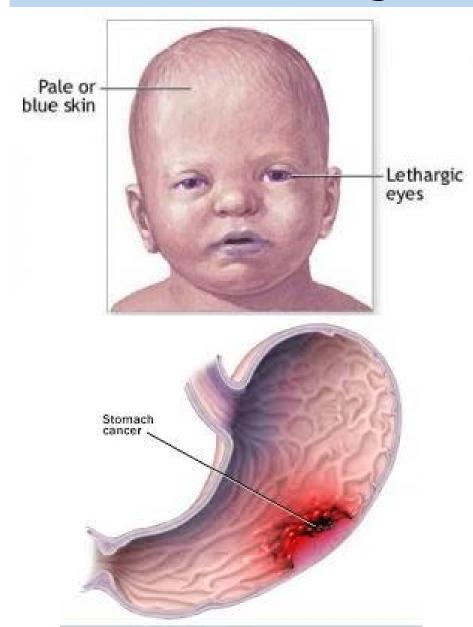
NO₃affect on Ecosystems





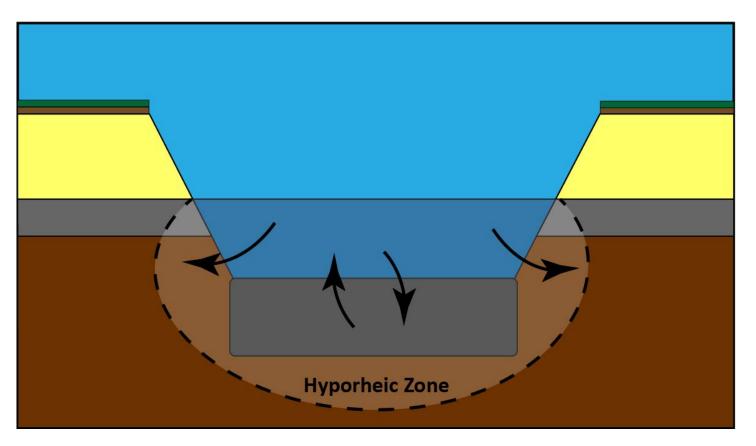
- Excess anthropogenic NO₃⁻ in surface waters are responsible for resulting in algal blooms leading hypoxia^[3-6]
- The 16,000 km² Gulf of Mexico hypoxic zone is primarily attributed to anthropogenic NO₃⁻ sourced from the Midwestern United States^[7-9]
- The Illinois River contributes 19% of the total NO₃⁻ load to the Gulf of Mexico hypoxic zone^[3-6]
- These hypoxic zones deteriorate ecosystems, leading to fish kills and benthic organism mortality^[9-10]

NO₃affect on Human Health



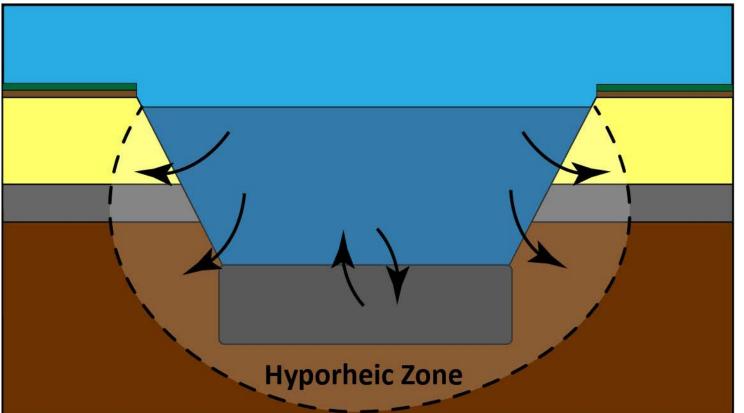
- National and Illinois EPA regard NO₃⁻ as a Primary National Drinking Standard^[11-12]
- NO₃⁻ limit for both National and Illinois EPA is 10 mgL⁻¹ as (NO₃⁻-N)^[11-12]
- Consumption of excess levels of NO₃⁻ can lead to:
 - Methemoglobinemia in infants and young children^[13]
 - The formation of nitrosamines attributed to gastric cancer^[14]

Hyporheic Zone (HZ) role in NO_3^- cycling



- HZ is an important area of surfacesubsurface interaction within streams & rivers^[15-22]
- Within the HZ denitrification can take place due to microorganisms who break NO_3^- down into $N_2O(g)$ and $N_2(g)^{[23-24]}$
 - This reaction is anaerobic or low dissolved oxygen (DO)^[23]
 - Requires high amounts of dissolved organic matter (DOM)^[23]
- Furthermore plant uptake can occur predominantly from benthic stream algae^[25-26]

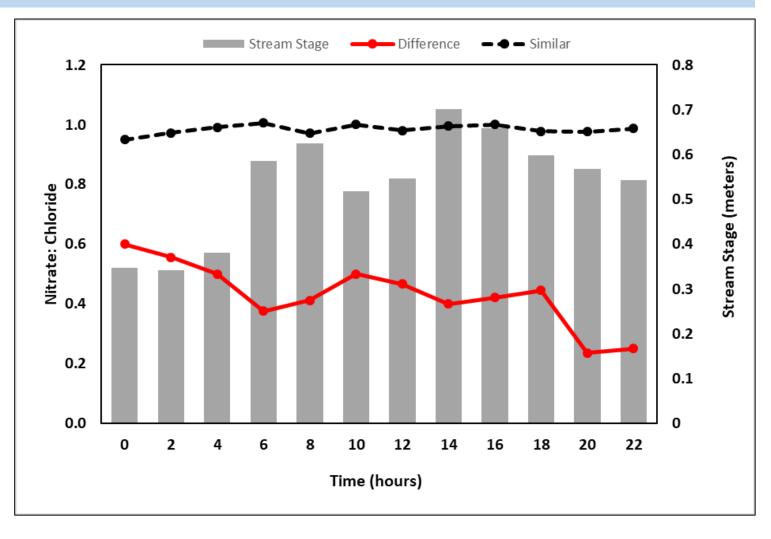
Storms Impact on NO₃⁻ in streams



- During storm events, stream stage increases as does the area of surfacesubsurface exchange and the volume of the HZ^[27-29]
 - DOC remains constant^[28]
 - DO increases^[27]
- However, it remains to be studied how NO₃⁻ behaves during elevated stream stage and whether or not stream banks play a crucial role in NO₃⁻ removal and retention.
- Using Chloride (Cl⁻), as a conservative tracer, we will compare its transport during storm events to that of NO₃⁻.

Objective

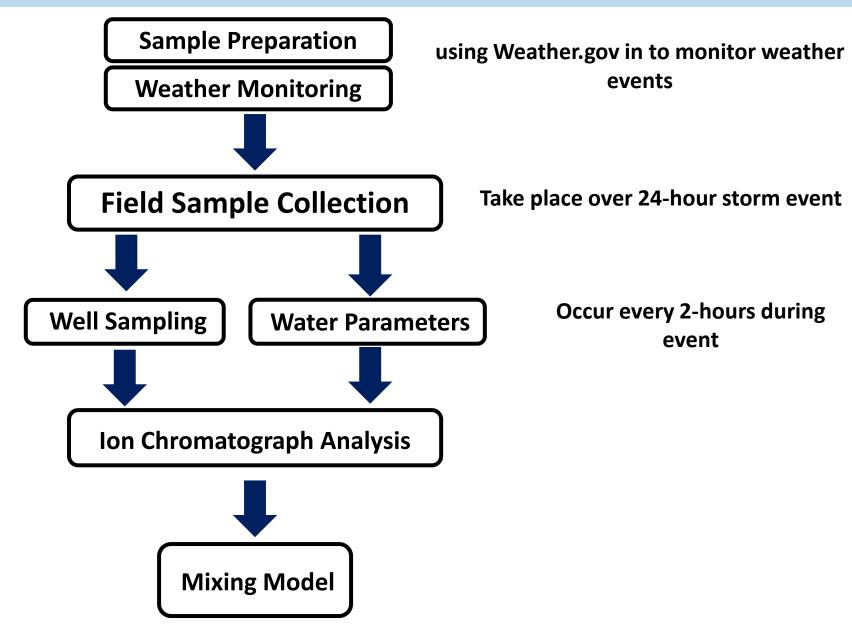
- If we measure that the behavior of NO₃⁻ and Cl⁻ concentrations during storm events are similar, then we can infer that transport during this time is conservative.
- If NO₃⁻ and Cl⁻ transport timing and amplitude of change differs may indicate that denitrification and uptake during storm events.



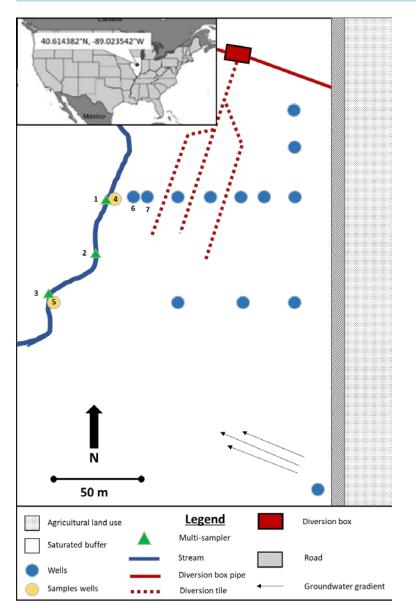
Questions

- How do concentrations of NO₃⁻ and Cl⁻ change in response to storm events within the stream, hyporheic zone, and bank storage?
- 2. Are NO₃⁻ and Cl⁻ transported similarly in a low-gradient system?
- 3. Are NO₃⁻ and Cl⁻ transported similarly during different storm events?

Methodology

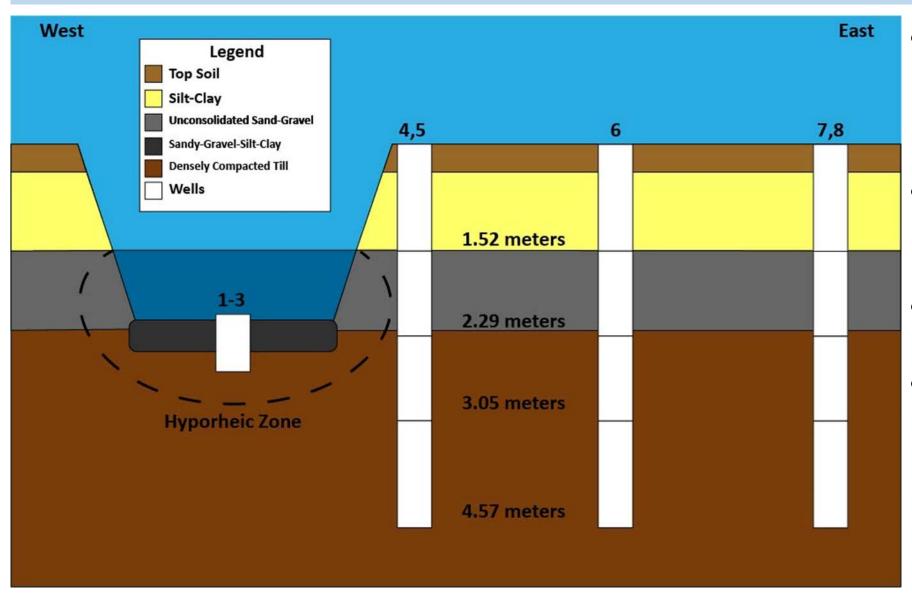


Study Site



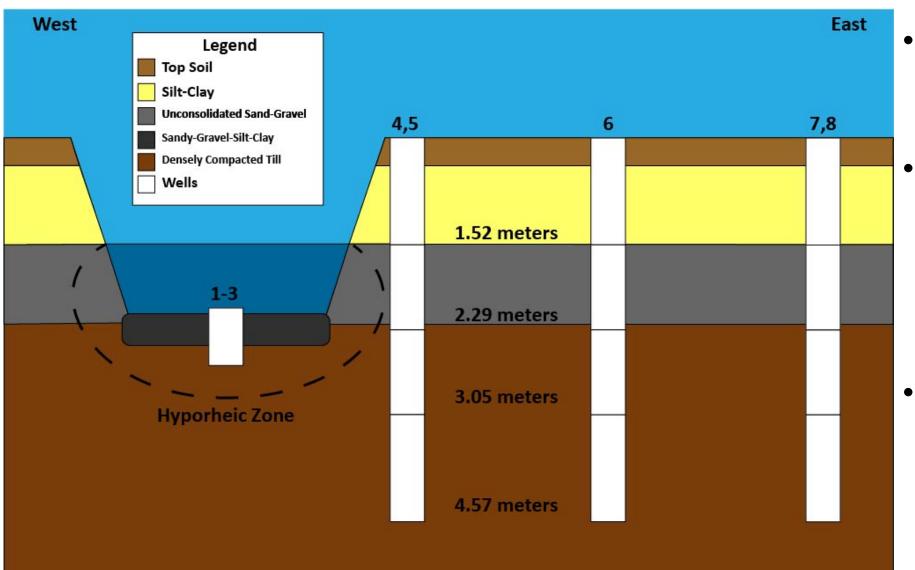
- T3 study site is a tributary of Evergreen Lake Watershed ELW) (105.45 km²)
 - ELW primary land cover is cultivated crops of 77.6%
 - ELW secondary land cover is developed land of 13.3 %
- T3 stream is a modified low-gradient stream fed by tile drainage.

Study Site



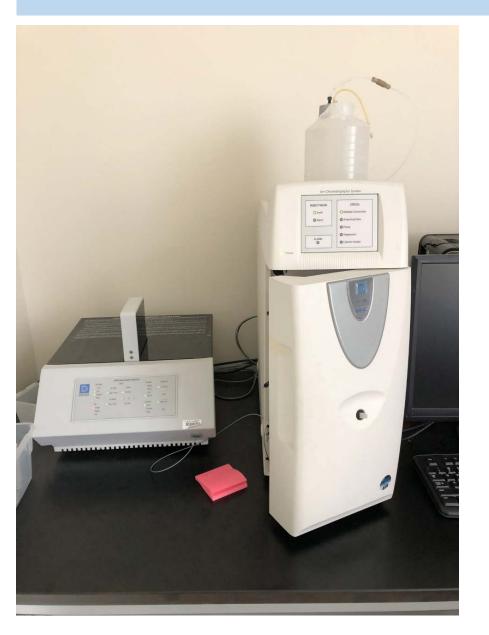
- Silt-Clay layer is 1 meter in depth composed of organicrich alluvium.
- Unconsolidated sandgravel layer ~ 1.25 meters in depth.
- Compacted glacial till sits beneath.
- Stream bed composed of sandy-gravel-siltclay layer ~50 cm in depth.

Field Sampling



- Sampling from 10, 30, and 50 cm depths in the stream and vertical HZ zone.
- 14 samples will be collected in total per rotation.
 - monitoring of pH, temperature, DO, and specific conductivity.
- Sample rotation will take place every 2 hours, with maximum monitoring duration lasting 24 hours.

Analysis



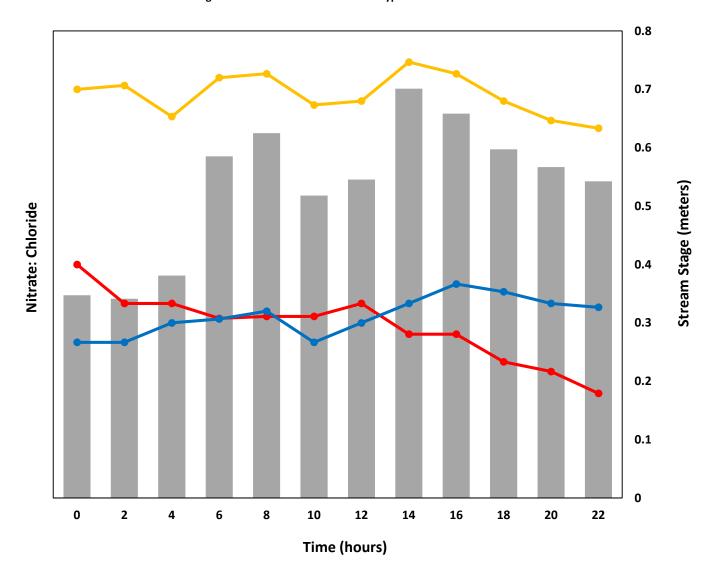
All samples will be analyzed in the Laboratory for Environmental Analysis (LEA) on a Dionex Ion Chromatograph, primarily measuring concentrations of NO_3^- and Cl^-

Will utilize a two-end member mixing model between Surface Water (SW) and Ground Waters (GW) to infer Stream Bank Concentrations of Cl^{-} ^[30] in order to determine NO_3^{-} mixing ^[30]

$$\% SW = \frac{(Cl_{HZ} - Cl_g)}{(Cl_S - Cl_g)}$$

 $NO_3 - N = \%SW * (N_s - N_g) + N_g$

Expected Results



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References

- 1. Panno, S. V., Kelly, W. R., Hackley, K. C., Hwang, H. H., & Martinsek, A. T. (2008). Sources and fate of nitrate in the Illinois River Basin, Illinois. Journal of Hydrology, 359(1–2), 174–188.
- 2. Kennedy, C. D., Bataille, C., Liu, Z., Ale, S., VanDeVelde, J., Roswell, C. R., ... & Bowen, G. J. (2012). Dynamics of nitrate and chloride during storm events in agricultural catchments with different subsurface drainage intensity (Indiana, USA). Journal of Hydrology, 466, 1-10.
- 3. Boulton, A. J., S. Findlay, P. Marmonier, E. H. Stanley, and H. M. Valett (1998), The functional significance of the hyporheic zone in streams and rivers, Annu. Rev. Ecol. Syst., 29, 59–81.
- 4. David MB, Gentry LE (2000) Anthropogenic inputs of nitrogen and phosphorus and riverine export for Illinois, USA. J Environ Qual 29:494–508.
- 5. David, M.B.; Wall, L.G.; Royer, T.V.; Tank, J.L. Denitrification and the nitrogen budget of a reservoir in an agricultural landscape. Ecol. Appl. 2006, 16, 2177–2190.
- 6. Keeney DR, Hatfield JL (2001) The nitrogen cycle: historical perspective, and current and potential future concerns. In: Follett R, Hatfield JL (eds) Nitrogen in the environment: sources, problems, and solutions. Elsevier, Amsterdam, pp 3–16
- 7. David MB, Drinkwater LE, McLsaac GF (2010) Sources of nitrate yields in the Mississippi River Basin. Journal of Environmental Quality 39:1657-1667.
- 8. Donner SD, Kucharik CJ, Foley JA (2004) Impact of changing land use practices on nitrate export by the Mississippi River. Global Biogeochemical Cycles 18.
- 9. Stets EG, Kelly VJ, Crawford CG (2015) Regional and temporal differences in nitrate trends discerned from long term water quality monitoring data. Journal of the American Water Resources Association 51:1394-1407.
- 10. Rabalais, N.N., R.E. Turner, and D. Scavia. 2002. Beyond science into policy: Gulf of Mexico hypoxia and the Mississippi River. BioScience 52:129–142.
- 11. Agency, U. S. E. P., & Water, O. (2018). 2018 Edition of the Drinking Water Standards and Health Advisories Tables.
- 12. Blagojevich, R. R. (2007). Maximum Levels for Contaminants in Public Water Supplies. 1–2.
- 13. Self, J. R., & Waskom, R. M. (1992). Nitrates in drinking water. Service in action; no. 0.517
- 14. Matiju, V., Cizinska, S., Krejci, J., Janoch, T., 1992. Biological water denitrification a review. Enzyme Microb. Technol. 14 (3), 170–183.
- 15. Brunke, M.; Gonser, T. The ecological significance of exchange processes between rivers and groundwater. Freshw. Biol. 1997, 37, 1–33.
- 16. Peterson, E. W., & Benning, C. (2013). Factors influencing nitrate within a low-gradient agricultural stream. Environmental Earth Sciences, 68(5), 1233–1245.
- 17. Cranswick, R.H.; Cook, P.G. Scales and magnitude of hyporheic, river-aquifer and bank storage exchange fluxes. Hydrol. Process. 2015, 29, 3084–3097
- 18. Harvey, J. W., Bohlke, J. K., Voytek, M. A., Scott, D., & Tobias, C. R. (2013). Hyporheic zone denitrification: Controls on effective reaction depth and contribution to whole-stream mass balance. Water Resources Research, 49, 6298–6316.
- 19. Hedin, L. O., von Fischer, J. C., Ostrom, N. E., Kennedy, B. P., Brown, M. G., & Robertson, G. P. (1998). Thermodynamic constraints on nitrogen transformations and other biogeochemical processes at soil-stream interfaces. Ecology, 79(2), 684–703.
- 20. Hinkle SR, Duff JH, Triska FJ, Laenen A, Gates EB, Bencala KE, Wentz DA, Silva SR (2001) Linking hyporheic flow and nitrogen cycling near the Willamette River, a large river in Oregon, USA. J Hydrol 244:157–180
- 21. Triska, F. J., Duff, J. H., & Avanzino, R. J. (1993). Patterns of hydrological exchange and nutrient transformation in the hyporheic zone of a gravel-bottom stream: Examining terrestrial-aquatic linkages. Freshwater Biology, 29(2), 259–274.
- 22. Squillace, P.J. Observed and simulated movement of bank-storage water. Ground Water 1996, 34, 121–134.
- 23. Mulholland, P. J., Hall Jr, R. O., Sobota, D. J., Dodds, W. K., Findlay, S. E., Grimm, N. B., ... & Ashkenas, L. R. (2009). Nitrate removal in stream ecosystems measured by 15N addition experiments: denitrification. Limnology and Oceanography, 54(3), 666-680.
- 24. Machefert, S. E., & Dise, N. B. (2004). Hydrological controls on denitrification in riparian ecosystems. Hydrology and Earth System Sciences, 8(4), 686–694.
- 25. Tischner, R. (2006). Nitrate uptake and reduction in plants. Journal of Crop Improvement, 15(2), 53–95.
- 26. Imsande, J., & Touraine, B. (1994). N demand and the regulation of nitrate uptake. Plant Physiology, 105(1), 3–7.
- 27. Malzone, J. M., Lowry, C. S., & Ward, A. S. (2016). Response of the hyporheic zone to transient groundwater fluctuations on the annual and storm event time scales. Water Resources Research, 52(7), 5301-5321.
- 28. Sawyer, A. H., Kaplan, L. A., Lazareva, O., & Michael, H. A. (2014). Hydrologic dynamics and geochemical responses within a floodplain aquifer and hyporheic zone during Hurricane Sandy. Water Resources Research, 50(6), 4877-4892.
- 29. Mueller Price, J., Bledsoe, B. P., & Baker, D. W. (2015). Influences of sudden changes in discharge and physical stream characteristics on transient storage and nitrate uptake in an urban stream. Hydrological Processes, 29(6), 1466–1479.
- 30. Peterson, E. W., & Hayden, K. M. (2018). Transport and fate of nitrate in the streambed of a low-gradient stream. Hydrology, 5(4), 34–38. https://doi.org/10.3390/hydrology5040055

Questions